

# **BOUMATIC LLC**

# Safety Data Sheet Tru-Blu Parlor Shine

# **SECTION 1: Identification**

1.1 Product identifier

Product name Tru-Blu Parlor Shine

Product number 8983016

1.2 Other means of identification

Alkaline Detergent Concentrate

1.4 Supplier's details

Name Boumatic LLC Address 2001 S. Stoughton

Madison, WI 53716

USA

Telephone 608-222-3484

email SDS@BouMatic.com

1.5 Emergency phone number(s)

24-Hour Emergency 1-800-255-3294 (U.S.)

001-813-248-0585 (International)

# **SECTION 2: Hazard identification**

# **General hazard statement**

The product is classified and labeled according to the Globally Harmonized System (GHS).

## 2.1 Classification of the substance or mixture

GHS classification in accordance with: (US) OSHA (29 CFR 1910.1200)

- Skin corrosion/irritation, Cat. 1A
- Eye damage/irritation, Cat. 1

# 2.2 GHS label elements, including precautionary statements

**Pictogram** 



Signal word Danger

Hazard statement(s)

H314 Causes severe skin burns and eye damage

H318 Causes serious eye damage

Precautionary statement(s)

P260 Do not breathe dust/fume/gas/mist/vapors/spray.

P264 Wash ... thoroughly after handling.

P280 Wear protective gloves/protective clothing/eye protection/face protection.

P280 Wear eve protection/face protection.

P301+P330+P331 IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.

P303+P361+P353 IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse

skin with water/shower.

P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing.

P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove

contact lenses if present and easy to do. Continue rinsing.

P310 Immediately call a POISON CENTER/doctor/...
P321 Specific treatment (see ... on this label).
P363 Wash contaminated clothing before reuse.

P405 Store locked up.

P501 Dispose of contents/container to ...

# **SECTION 3: Composition/information on ingredients**

# 3.2 Mixtures

#### **Hazardous components**

#### 1. Sodium hydroxide

 Concentration
 Not specified

 EC no.
 215-185-5

 CAS no.
 1310-73-2

 Index no.
 011-002-00-6

- Skin corrosion/irritation, Cat. 1A

H314 Causes severe skin burns and eye damage

#### 2. Potassium hydroxide

 Concentration
 Not specified

 EC no.
 215-181-3

 CAS no.
 1310-58-3

 Index no.
 019-002-00-8

- Skin corrosion/irritation, Cat. 1A

- Acute toxicity, oral, Cat. 4

H302 Harmful if swallowed

H314 Causes severe skin burns and eye damage

# 3. Sodium hypochlorite solution (4-6% cl2)

 Concentration
 Not specified

 EC no.
 231-668-3

 CAS no.
 7681-52-9

 Index no.
 017-011-00-1

- Skin corrosion/irritation, Cat. 1B

- Hazardous to the aquatic environment, short-term (acute), Cat. 1

H314 Causes severe skin burns and eye damage

H400 Very toxic to aquatic life

## 4. 2-Propenoic acid, telomer with sodium sulfite (1:1), sodium salt

Concentration Not specified CAS no. 68479-09-4

# 5. Dimethyldodecylamine-n-oxide

Concentration Not specified CAS no. 1643-20-5

# **SECTION 4: First-aid measures**

## 4.1 Description of necessary first-aid measures

General advice Immediately remove any clothing soiled by the product.

If inhaled If breathed in, move person into fresh air. If not breathing, give artificial

respiration. Consult a physician.

In case of skin contact Wash off with soap and plenty of water. Consult a physician.

In case of eye contact Rinse thoroughly with plenty of water for at least 15 minutes and consult a

physician.

If swallowed Drink copious amounts of water and provide fresh air. Consult a physician.

Personal protective equipment for first-aid responders

No further relevant information available.

#### 4.2 Most important symptoms/effects, acute and delayed

No further relevant information available.

#### 4.3 Indication of immediate medical attention and special treatment needed, if necessary

No further relevant information available.

# **SECTION 5: Fire-fighting measures**

#### 5.1 Suitable extinguishing media

Use water spray, alcohol-resistant foam, dry chemical or carbon dioxide.

# 5.2 Specific hazards arising from the chemical

No further relevant information available.

#### 5.3 Special protective actions for fire-fighters

No special measures required.

#### Further information

No further relevant information available.

#### SECTION 6: Accidental release measures

# 6.1 Personal precautions, protective equipment and emergency procedures

Ensure adequate ventilation. Use personal protective equipment. For personal protection see section 8.

#### 6.2 Environmental precautions

Prevent further leakage or spillage if safe to do so. Do not let product enter drains. Discharge into the environment must be avoided.

# 6.3 Methods and materials for containment and cleaning up

Soak up with inert absorbent material (e.g. sand, silica gel). Keep in suitable, closed containers for disposal.

#### Reference to other sections

See Section 7 for information on safe handling. See Section 8 for information on personal protection equipment. See Section 13 for disposal information.

# **SECTION 7: Handling and storage**

#### 7.1 Precautions for safe handling

Ensure adequate ventilation. Wash hands before breaks and immediately after handling the product. Avoid contact with skin, eyes and clothing. Avoid ingestion and inhalation. For precautions see section 2.2.

#### 7.2 Conditions for safe storage, including any incompatibilities

Keep container tightly closed in a dry and well-ventilated place.

#### Specific end use(s)

No further relevant information available.

# **SECTION 8: Exposure controls/personal protection**

### 8.1 Control parameters

#### 1. Sodium hydroxide (CAS: 1310-73-2)

PEL (Inhalation): 2 mg/m3; USA (OSHA) OSHA Annotated Table Z-1, www.osha.gov

PEL (Inhalation): (C) 2 mg/m3; USA (Cal/OSHA)

OSHA Annotated Table Z-1, www.osha.gov

REL (Inhalation): (C) 2 mg/m3; USA (NIOSH) OSHA Annotated Table Z-1, www.osha.gov

TLV® (Inhalation): (C) 2 mg/m3; USA (ACGIH) OSHA Annotated Table Z-1, www.osha.gov

# 2. Potassium hydroxide (CAS: 1310-58-3 EC: 215-181-3)

PEL-C (Inhalation): 2 mg/m3; USA (ACGIH)

Upper Respiratory Tract irritation, Eye irritation, Skin irritation

PEL-C (Inhalation): 2 mg/m3; USA (ACGIH)

Upper Respiratory Tract irritation, Eye irritation, Skin irritation

PEL-C (Inhalation): 2 mg/m3; USA (NIOSH)
PEL-C (Inhalation): 2 mg/m3; USA (Cal/OSHA)

### 8.2 Appropriate engineering controls

Handle in accordance with good industrial hygiene and safety practice. Wash hands before breaks and at the end of workday.

## 8.3 Individual protection measures, such as personal protective equipment (PPE)

# Eye/face protection

Tightly fitting safety goggles. Faceshield (8-inch minimum). Use equipment for eye protection tested and approved under appropriate government standards such as NIOSH (US) or EN 166(EU).

# Skin protection

Handle with gloves. Gloves must be inspected prior to use. Use proper glove removal technique (without touching glove's outer surface) to avoid skin contact with this product. Dispose of contaminated gloves after use in accordance with applicable laws and good laboratory practices. Wash and dry hands. The glove material has to be impermeable and resistant to the product/the substance/the preparation. Due to missing tests, no recommendation to the glove material can be given for the product/the preparation/the chemical mixutre. Selection of glove material on consideration of penetration times, rates of diffusion and degredation.

# **Body protection**

Choose body protection in relation to its type, to the concentration and amount of dangerous substances, and to the specific work-place., The type of protective equipment must be selected according to the concentration and amount of the dangerous substance at the specific workplace.

# Respiratory protection

In case of brief exposure or low pollution, use respiratory filter device. In case of intensive or longer exposure use respiratory protective device that is independent of circulating air.

#### Thermal hazards

No data available

#### **Environmental exposure controls**

Prevent further leakage or spillage if safe to do so. Do not let product enter drains. Discharge into the environment must be avoided.

# **SECTION 9: Physical and chemical properties**

Information on basic physical and chemical properties

Appearance/form (physical state, color, etc.)

Odor

Odor threshold

рΗ

Melting point/freezing point

Initial boiling point and boiling range

Flash point Evaporation rate

Flammability (solid, gas)
Upper/lower flammability limits
Upper/lower explosive limits

Vapor pressure Vapor density Relative density Solubility(ies)

Partition coefficient: n-octanol/water

Auto-ignition temperature Decomposition temperature

Viscosity

Explosive properties Oxidizing properties Liquid. Yellow. Chlorine

Not determined Not determined 100 C (212 F) Not applicable Not determined Not applicable Not applicable Not determined

at 20 C (68 F): 23 hPa (17 mm HG)

Not determined Not determined

Not miscible or difficult to mix

Not determined

Product is not self-igniting

Not determined Not determined

Product does not present an explosion hazard.

# **SECTION 10: Stability and reactivity**

### 10.1 Reactivity

None under normal use conditions.

#### 10.2 Chemical stability

No data available

#### 10.3 Possibility of hazardous reactions

No dangerous reactions known.

#### 10.4 Conditions to avoid

No further relevant information available.

## 10.5 Incompatible materials

----

Sodium hydroxide: Caustic soda reacts with all the mineral acids to form the corresponding salts. It also reacts with weak-acid gases, such as hydrogen sulfide, sulfur dioxide, and carbon dioxide. Caustic soda reacts with amphoteric metals (Al, Zn, Sn) and their oxides to form complex anions such as AlO2(-), ZnO2(-2), SNO2(-2), and H2 (or H2O with oxides). All organic acids also react with sodium hydroxide to form soluble salts. Another common reaction of caustic soda is dehydrochlorination.

----

Potassium hydroxide: Nitro compounds, Organic materials, Magnesium, Copper, Water, reacts violently with:, Metals, Light metals, Contact with aluminum, tin and zinc liberates hydrogen gas. Contact with nitromethane and other similar nitro compounds causes formation of shock-sensitive salts., vigorous reaction with:, Alkali metals, Halogens, Azides, Anhydrides

# 10.6 Hazardous decomposition products

----

Sodium hydroxide: Sodium oxides

----

Potassium hydroxide: Other decomposition products - No data available

Hazardous decomposition products formed under fire conditions. - Potassium oxides

In the event of fire: see section 5

# **SECTION 11: Toxicological information**

#### Information on toxicological effects

#### **Acute toxicity**

LD/LC50 values that are relevant for classification:

1310-73-2 Sodium Hydroxide, Oral [LD50] 2000 mg/kg (Rat)

1310-58-3 Potassium Hyrdoxide, Oral [LD50] 273 mg/kg (Rat)

#### Skin corrosion/irritation

Causes severe skin burns.

# Serious eye damage/irritation

Risk of serious damage to eyes.

# Respiratory or skin sensitization

Strong caustic effect on skin and mucous membranes.

#### Germ cell mutagenicity

No data available

## Carcinogenicity

IARC (International Agency for Research on Cancer): None of the ingredients listed.

NTP (National Toxicity Program): None of the ingredients listed.

OSHA-Ca (Occupational Safety & Health Administration: None of the ingredients listed.

# Reproductive toxicity

No data available

# Summary of evaluation of the CMR properties

Carcinogenic catagories: IARC (International Agency for Research on Cancer), NTP (National Toxicology Program), OSHA-Ca (Occupational Safety and Health Administration): None of the ingredients listed.

# STOT-single exposure

No data available

# STOT-repeated exposure

No data available

## **Aspiration hazard**

May be harmful if swallowed and enters airways

# **Additional information**

To the best of our knowledge, the chemical, physical, and toxicological properties have not been thoroughly investigated.

# **SECTION 12: Ecological information**

## **Toxicity**

Aquatic toxicity: No further relevant information available.

## Persistence and degradability

No data available on product

#### Bioaccumulative potential

No data available on product

#### Mobility in soil

No data available on product.

#### Results of PBT and vPvB assessment

PBT/vPvB assessment not available as chemical safety assessment not required/not conducted

#### Other adverse effects

No further relevant data available.

# **SECTION 13: Disposal considerations**

#### Disposal of the product

Dispose of contents/ container in accordance with the local/regional/national/international regulations. Non Household Setting: Products covered by this SDS, in their original form, when disposed as waste, are considered non hazardous waste according to Federal RCRA regulations (40 CFR 261). Disposal should be in accordance with local, state and federal regulations.

# Disposal of contaminated packaging

Disposal must be made according to official regulations.

## Waste treatment

Disposal must be made according to official regulations.

#### Sewage disposal

Disposal must be made according to official regulations.

#### Other disposal recommendations

Disposal must be made according to official regulations.

# **SECTION 14: Transport information**

DOT (US)

UN Number: UN3266

Class: 8

Packing Group: II

Proper Shipping Name: UN3266, Corrosive Liquid, Basic, Inorganic, (Sodium Hypochlorite, Sodium Hydroxide,

Potassium Hydroxide)

Reportable quantity (RQ): N/A

Marine pollutant: No

Poison inhalation hazard: N/A

**IMDG** 

UN Number: UN3266

Class: 8

Packing Group: II EMS Number: F-A-S-B

Proper Shipping Name: UN3266, Corrosive Liquid, Basic, Inorganic, (Sodium Hypochlorite, Sodium Hydroxide,

Potassium Hydroxide)

**IATA** 

UN Number: UN32266

Class: 8

Packing Group: II

Proper Shipping Name: UN3266, Corrosive Liquid, Basic, Inorganic, (Sodium Hypochlorite, Sodium Hydroxide,

Potassium Hydroxide)

# **SECTION 15: Regulatory information**

# 15.1 Safety, health and environmental regulations specific for the product in question

# **Massachusetts Right To Know Components**

Chemical name: Sodium hydroxide

CAS number: 1310-73-2

#### **New Jersey Right To Know Components**

Common name: Sodium hydroxide

CAS number: 1310-73-2

# Pennsylvania Right To Know Components

Chemical name: Sodium hydroxide

CAS number: 1310-73-2

## **SARA 302 Components**

No chemicals in this material are subject to the reporting requirements of SARA Title III, Section 302.

#### **SARA 313 Components**

This material does not contain any chemical components with known CAS numbers that exceed the threshold (De Minimis) reporting levels established by SARA Title III, Section 313.

#### SARA 311/312 Hazards

Acute Health Hazard

## California Prop. 65 Components

This product does not contain any chemicals known to State of California to cause cancer, birth defects, or any other reproductive harm.

Version: 1.0, Date of issue: 2019-08-22, p. 9 of 11

## **Massachusetts Right To Know Components**

Potassium hydroxide CAS-No. 1310-58-3

# **New Jersey Right To Know Components**

Potassium hydroxide CAS-No. 1310-58-3

# Pennsylvania Right To Know Components

Potassium hydroxide CAS-No. 1310-58-3

# California Prop. 65 Components

This product does not contain any chemicals known to State of California to cause cancer, birth defects, or any other reproductive harm.

# **Massachusetts Right To Know Components**

Chemical name: Sodium hypochlorite

CAS number: 7681-52-9

# **New Jersey Right To Know Components**

Common name: Sodium hypochlorite

CAS number: 7681-52-9

# Pennsylvania Right To Know Components

Chemical name: Hypochlorous acid, sodium salt

CAS number: 7681-52-9

#### 15.2 Chemical Safety Assessment

Keep out of reach of children. Read label before use. Wear protective gloves/protective clothing/eye protection/face protection. Use personal protective equipment as required.

# **HMIS Rating**

Tru-Blu Parlor Shine	
HEALTH	3
FLAMMABILITY	0
PHYSICAL HAZARD	0
PERSONAL PROTECTION	

# **NFPA Rating**



# **SECTION 16: Other information**

# 16.1 Further information/disclaimer

DISCLAIMER: The information above is believed to be accurate and represents the best information currently available to us. However, we make no warranty of merchantability or any other warranty, express or implied, with respect to such information, and we assume no liability resulting from its use. Users should make their own investigation to determine the suitability of information for their particular purposes. In no event shall BouMatic be liable for any claims, losses, or damages of any third party or for lost profits or any special, indirect, incidental, consequential or exemplary damages, whatsoever arising, even if BouMatic has been advised of the possibility of such damages.

#### 16.2 Preparation information

This information is based on our present knowledge. However, this shall not constitute a guarantee for any specific product features and shall not establish a legally valid contractual relationship.